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From this definition, an acid donates Hl +. Hl + has no electrons and neutrons .It contains only a proton. Examples. I. From the equation: HCl(aq) + H 2 O(l) === H 3 O+(aq) + Cl-(aq) (a)(i)For the forward reaction from left to right, H 2 O gains a proton to form H O+ and thus H 2 O is a proton acceptor . It is a Bronsted-Lowry base

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